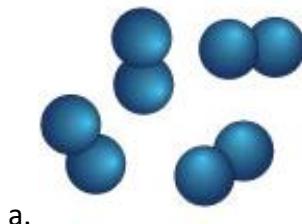
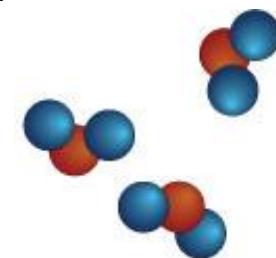


8-sinf namuna

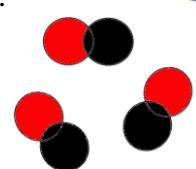
1. Oddiy modda berilgan qartorni ko'rsating.



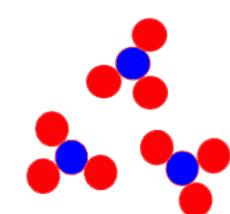
a.



b.

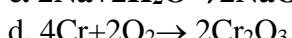
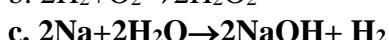
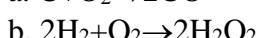
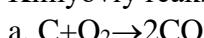


c.



d.

2. Kimyoviy reaksiya tenglamasini to'g'ri tenglangan qartorni ko'rsating.



3. 3,2 gramm kislород гази тарқибидаги молекулалар сонини исобланг.

a.  $6,02 \cdot 10^{22}$

b.  $12,04 \cdot 10^{22}$

c.  $12,04 \cdot 10^{23}$

d.  $6,02 \cdot 10^{23}$

4. 100 gramm 50 % li eritma tayyorlash uchun qancha tuz kerak bo'ladi.

a. **50 g**

b. 180 g

c. 10 g

d. 90 g

5. 500 gramm eritmaning hajmi  $400 \text{ cm}^3$  bo'lsa eritmaning zichligini hisoblang.

a. **1,25 g/cm<sup>3</sup>**

b. 1 g/cm<sup>3</sup>

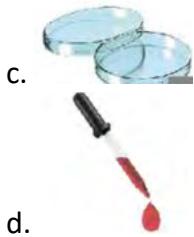
- c.  $0,8 \text{ cm}^3/\text{g}$   
d.  $1,2 \text{ g/cm}^3$
6. Vodorod qaysi gaz bilan xona haroratidayoq birikadi.  
**a.  $\text{F}_2$**  b.  $\text{Cl}_2$  c.  $\text{N}_2$  d.  $\text{Na}$
7. 32 gramm natriy gidroksid sulfat kislota bilan reaksiyaga kirishib necha mol o'rta tuz hosil qiladi. (reaksiya natijasida faqat o'rta tuz hosil bo'lgan)  
a. 0,2 mol  
b. 0,1 mol  
c. 0,3 mol  
**d. 0,4 mol**
8. 3,4 gramm vodorod peroksidning to'liq parchalanganda necha gramm kislород hosil bo'ladi.  
a. 3,2 gramm  
**b. 1,6 gramm**  
c. 0,32 gramm  
d. 0,16 gramm
9. Amfoter oksid berilgan qatorni ko'rsating.  
**a.  $\text{ZnO}$**  b.  $\text{CaO}$  c.  $\text{CO}$  d.  $\text{CO}_2$
10. 50 % kislород va 50% ozondan iborat gazlar aralashmasining geliyga nisbatan zichligini aniqlang.  
**a. 10**  
b. 12  
c. 40  
d. 30
11. 2 valentli noma'lum 0,45 gramm X metal 0,8 gramm kislород bilan to'liq reaksiyaga kirishdi. Nomalum metalni aniqlang.  
**a. Be**  
b. Mg  
c. Ar  
d. Al
12. Alyuminiy sulfat formulasini to'g'ri berilgan qatorni ko'rsating.  
a.  $\text{Al}(\text{HSO}_4)_3$   
**b.  $\text{Al}_2(\text{SO}_4)_3$**   
c.  $\text{Al}_2(\text{HSO}_4)_3$   
d.  $\text{Al}_3\text{HSO}_4$
13. Quyidagilardan qaysi biri qisqich hisoblanadi.



a.



b.



- c.
- d.
14. Ca va Ag dan iborat 10 gramm aralshma yetarli miqdorda xlorid kislota bilan ishlanganda  $3,36 \text{ dm}^3$  (n.sh. da) gaz ajralib chiqdi. Aralashma tarkibidagi kumushning massasini hisoblang.
- a. **4 g**
  - b. 6 g
  - c. 3 g
  - d. 7 g
15. Flagiston nazariyasi muallifi kim?
- a. Shtall
  - b. Lamonosov
  - c. Butlerov
  - d. Lavuaze